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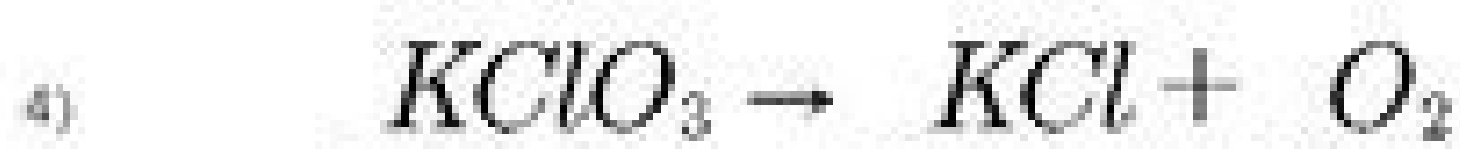
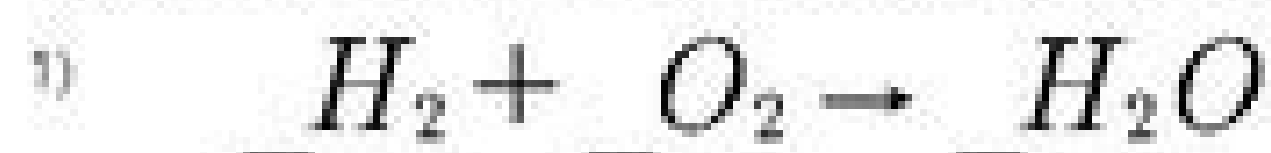

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Balancing Equations Practice

Instructions:

Balance each equation and identify the type of chemical reaction.



Balancing Chemical Equations Worksheet 1

1. $H_2 + O_2 \rightarrow H_2O$
2. $Na + Cl_2 \rightarrow NaCl$
3. $N_2O_4 \rightarrow NO_2$
4. $Mg + O_2 \rightarrow MgO$
5. $H_2O_2 \rightarrow H_2O + O_2$
6. $Ca + N_2 \rightarrow Ca_3N_2$
7. $Li + F_2 \rightarrow LiF$
8. $Mg + N_2 \rightarrow Mg_3N_2$
9. $NH_3 \rightarrow N_2 + H_2$
10. $HCl \rightarrow H_2 + Cl_2$
11. $Nb \rightarrow N_2 + I_2$
12. $HI \rightarrow H_2 + I_2$
13. $Al + O_2 \rightarrow Al_2O_3$
14. $K + S \rightarrow K_2S$
15. $Li + Cl_2 \rightarrow LiCl$
16. $HgO \rightarrow Hg + O_2$
17. $Al + I_2 \rightarrow AlI_3$
18. $Al + S \rightarrow Al_2S_3$
19. $Na + H_2O \rightarrow NaOH + H_2$
20. $NO + O_2 \rightarrow NO_2$
21. $Na_2O + H_2O \rightarrow NaOH$
22. $CH_4 + O_2 \rightarrow CO_2 + H_2O$
23. $PbO + C \rightarrow Pb + CO_2$



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